

Phase separation of VO₂ and SiO₂ on SiO₂-Coated float glass yields robust thermochromic coating with unrivalled optical properties

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ABSTRACT

Vanadium dioxide displays thermochromic properties based on its structural phase transition from monoclinic VO₂ (M) to rutile VO₂ (R) and *vice versa*, and the accompanying reversible metal-insulator transition. We developed a single layer coating comprising VO₂ (M) and SiO₂. We applied the coating from an alcoholic solution comprising vanadium(IV) oxalate complex and pre-oligomerized tetra ethoxy silane to SiO₂-coated float glass using dip coating, and thermally annealed the dried xerocoat in a two-step process. The addition of SiO₂ as coating matrix resulted in non-scattering coatings with low surface roughness and random distribution of VO₂ nanodomains (<200 nm). Furthermore, the formation of the coating, comprising a phase separation yielding SiO₂ and VO₂ nanodomains during the thermal anneal, was studied in detail. The coating displays unrivalled optical properties, combining high visible light transmission $T_{vis} > 60\%$ and large solar modulation $\Delta T_{sol} \geq 10\%$. When applied in insulating glass units, the coating has a positive impact on energy savings for heating and cooling of buildings in intermediate climates, which we demonstrated through building energy simulations. For a typical house in the Netherlands, energy savings up to 24% were obtained. In addition, we demonstrate a coating stability comparable to current energy-efficient window coatings during processing into and in insulating glass units through (accelerated) life time tests.

1. Introduction

Thermochromic coatings are of interest for application in smart windows since they change their light transmission at a critical switching temperature T_0 [1–3]. The thermochromic material that is most promising for use in window coatings is monoclinic vanadium dioxide – VO₂ (M) [3–5]. VO₂ (M) undergoes a structural phase transition to rutile VO₂ (R) at $T_0 = 68$ °C, which is reversible and accompanied by a metal-insulator transition (MIT). Dopants, such as W, can reduce T_0 to values between 15 °C and 30 °C, which is ideal for smart windows. The MIT does not only cause a change in the electrical but also in the optical properties of the coating [3]. When switching from VO₂ (M) to VO₂ (R), the transmission of solar infrared (IR) light decreases.

This leads to a reduction in solar heat gain at high outside temperatures, keeps buildings cooler in summer and reduces the energy consumption and costs for the use of air-conditioning systems. The benefits of using VO₂ in window coatings are that the switch in optical properties occurs mainly in the solar IR region and is therefore largely invisible [6], that the change in light transmission between the high temperature and low temperature phase can be tuned, e.g. by varying coating thickness [3] or composition [7], and that the switching temperature can be adjusted by addition of dopants [8]. The switching temperature is defined either as the temperature at start of the structural phase transition (T_0 ; thermodynamically relevant temperature) [9], or as the temperature of the heat flow optimum (T_{switch} ; commonly reported in application oriented publications) [3]. Many studies have been published investigating VO₂ in

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single [7,10,11] or multi-layer coatings [12,13], nanoporous [14–17] or nanocomposite single layers [7,16–19], incorporated in multilayer antireflective coating stacks [20–23], or as pigment material in nanocomposite polymer films [24–27]. Major goal of these studies was to optimize the optical properties of VO₂ by combining high visible transmission (T_{vis}) with a high solar modulation (ΔT_{sol}), where

$$T_{vis,sol} = \frac{\int \phi_{vis,sol}(\lambda)T(\lambda)d\lambda}{\int \phi_{vis,sol}(\lambda)d\lambda} \quad (1)$$

$$\Delta T_{sol} = T_{sol,cold} - T_{sol,hot} \quad (2)$$

with $T(\lambda)$ as the transmittance at wavelength λ , $\phi_{vis}(\lambda)$ as the photopic spectral sensitivity of human vision, and $\phi_{sol}(\lambda)$ as the AM 1.5 solar irradiance spectrum. Nevertheless, due to various drawbacks of the different types of coatings, no coating suitable for industrial scale implementation in smart windows has been reported to date. Single and multilayer coatings of pure VO₂ suffer from the non-linear change of refractive index (n) upon phase transition [23]. For such coatings the refractive index in wavelength regions below 2 μm is lower for the high temperature rutile phase VO₂ (R) than the low temperature monoclinic phase VO₂ (M). Only in high wavelength regions above 2 μm does the refractive index increase while changing from VO₂ (M) to VO₂ (R). This leads to reduced reflection at high temperatures, and therefore increased transmission especially in the wavelength region between 500 and 900 nm [23]. Therefore, even for thick coatings, which display low T_{vis} , only moderate values for ΔT_{sol} are reached [3]. These issues can be circumvented by e.g. including the VO₂ layer in an antireflective coating stack. Here, by carefully adjusting the refractive index of all layers, high ΔT_{sol} values of up to 21.8% can be reached at T_{vis} ranging from 45 to 55% [20–23]. For commercial application these systems are less relevant due to their complexity with multiple coating layers, which leads to high production costs. Another possibility to tune the refractive index of the VO₂ layer is by creating nanoporous or nanocomposite coatings. The introduction of nanometer sized air pockets [14–17] or low refractive index material as matrix for VO₂ domains [7,16–19] can lead to a better balance between T_{vis} and ΔT_{sol} . Most of the methods to prepare these types of coatings, e.g. via hydrothermal growth of coatings on glass [14], via freeze-drying [15] or coatings comprising porous or core-shell nanoparticles [16,17], are not feasible on commercial scale. Other methods that are applicable for large scale production only show large ΔT_{sol} at low $T_{vis} < 50\%$ [7,18,19]. The only systems reported to date that combine high $T_{vis} > 60\%$ with large $\Delta T_{sol} \geq 10\%$ in processes viable for large scale production are nanocomposite polymer films [24–27], which can be used as interlayers in laminated glass. The introduction of VO₂ nanopigments in several micrometer thick polymer films can lead to a combination of high T_{vis} and large ΔT_{sol} values. However, most of these nanocomposite films exhibit a broad hysteresis and transition gradient [24–27], which negatively affect the energy savings potential of smart windows [28]. Recently, Ciu et al. [3] reviewed the developments on VO₂-based thermochromic coatings and films, and concluded that for application in smart windows it is desired to combine high $T_{vis} (>60\%)$ with high $\Delta T_{sol} (\geq 10\%)$ which has not yet been achieved with single-layer thermochromic coatings. One of the most promising approaches using an industrially relevant technique was reported by Schläfer et al. [7]. Here composite coatings, integrating small VO₂ domains in an SiO₂ matrix via a sol-gel coating technique, were investigated. They reported an optimized balance between T_{vis} and ΔT_{sol} and showed via theoretical calculations on different compositions and optical constants the potential of these composite coatings. Even though the theoretical calculations projected that it should be possible to combine high $\Delta T_{sol} \geq 10\%$ with $T_{vis} > 60\%$ using this system, only $T_{vis} < 53\%$ at $\Delta T_{sol} \geq 10\%$ was achieved in composite coatings.

To progress the development of thermochromic VO₂ coatings from a promising concept to successful market introduction in energy-efficient glazing, it is important to develop a robust coating that combines high

$T_{vis} > 60\%$ with high $\Delta T_{sol} \geq 10\%$ in a production process that is feasible at commercial scale. Furthermore, the coating has to be stable under processing conditions for integration of coated glass plates in an insulating glass unit (IGU), and for more than 10 years within the final IGU installed in a building. We addressed all these challenges and report the first robust nanocomposite single layer VO₂ coating combining high $T_{vis} > 60\%$ with large $\Delta T_{sol} \geq 10\%$ applied in a cost efficient and commercially feasible process. We developed a single layer coating comprising VO₂ (M) and SiO₂. We applied the coating from an alcoholic solution comprising vanadium(IV) oxalate complex and pre-oligomerized tetraethoxy silane to SiO₂-coated float glass using dip coating, and thermally annealed the xerocoat in a two-step process. The formation of the coating, comprising a phase separation during the thermal anneal yielding SiO₂ and VO₂ nanodomains, was studied in detail. We investigated the influence of the VO₂ (M) concentration and thickness of the nanocomposite coating on T_{vis} and ΔT_{sol} of the coated glass. Furthermore, we analyzed the stability of a single side coated glass plate for common processing conditions of glass plates into an IGU and we performed accelerated lifetime tests according to standards for IGUs coated on the inside of the outer glass pane (layer 2, Fig. 1). Additionally, we studied the energy savings potential of our high performance coatings in an average Dutch building using building energy simulations.

2. Materials and methods

2.1. Materials

Vanadium(V) oxide (V₂O₅, 99.6%), oxalic acid (98%) and 2-propanol (99.8%) were purchased from Sigma-Aldrich. Acetic acid (99.9%) was purchased from Alfa Aesar. Tetraethoxysilane (TEOS, 98%) was purchased from Acros Organics. Nitric acid (65%) was purchased from Fisher Chemicals. Nitrogen gas (N₂, HiQ 6.0) was purchased from Linde Gas. All chemical were used as received without further purification. Vanadyl oxalate solution and silica sol were prepared via previously reported procedures [9]. As glass substrates 10 × 10 cm² Pilkington Optiwhite™ glass plates of 4 mm thickness were used. Prior to application of the thermochromic coating the glass substrate was cleaned and a silica barrier coating was applied according to a previously reported procedure [9]. All coating experiments were performed at a relative humidity below 35% and a temperature between 19 °C and 25 °C.

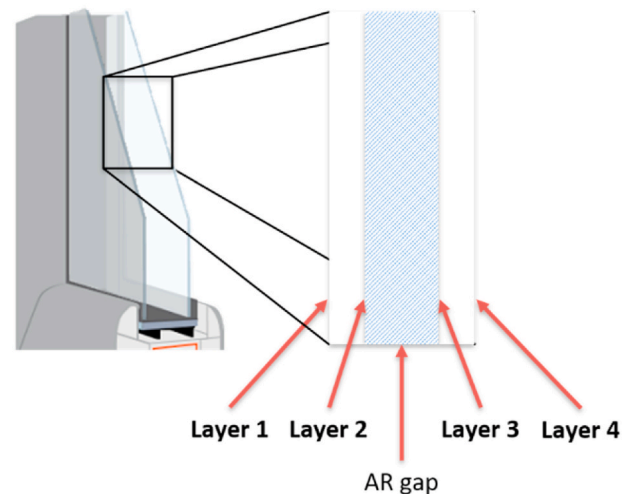


Fig. 1. IGU with typical labeling of the individual glass surfaces. Layer 1 is in contact with the outside environment and layer 4 with the inside environment. Our thermochromic coating is applied on layer 2.

2.2. Preparation of thermochromic coatings

One side of the barrier coated glass substrate was masked using d-c-fix® self-adhesive foil. Subsequently, the thermochromic coating was applied to the non-covered, non-tin side using dip coating. The applied liquid coating formulation consisted of a mixture of vanadyl oxalate solution and silica sol at an appropriate ratio of SiO₂ to VO₂ to reach the desired composition (Table 1). The xerogel coating was applied to the glass substrates at withdrawal speeds in the range of 1.0–5.0 mm s⁻¹ to achieve the desired thickness. After drying for 5 min under ambient conditions, the adhesive foil was removed. The non-coated side was then placed onto a 6 inch silica wafer, and annealed in a Jipelec Jetfirst PV Rapid Thermal Processor, with the coated side facing the IR radiators. The coated glass plates were first heated to 270 °C in air for 480 s, and subsequently in N₂ at 450 °C for 30 s.

2.3. Building energy simulations

Building energy simulations have been performed using Energy Plus version 9.2.0 [29]. The selected simulation period was one year with data points gathered monthly. For a typical Dutch building (Fig. S1) one half of a three story duplex house was chosen with a floor area of 5 × 10 m² and a ceiling height of 2.6 m (Fig. 2). With a 45° roof pitch starting at the second floor level, a total living space of 130 m² was obtained according to standard definitions. All walls were designed facing one cardinal direction with the western wall not in contact to the outside environment. In agreement with typical window surface values for duplex houses in the Netherlands, we selected a total window surface of 35 m². The outer walls were designed according to Dutch regulations to reach a thermal insulation value (Rc-value) of at least 4.5 m² K W⁻¹. For detailed description of wall material and built up see supporting information (Fig. S2, Tables S1 and S2). Illuminance level and lighting loads were adjusted to reach average annual energy consumption values for lighting for a 4 person household in the Netherlands. All other parameters were used as reported in our previous building energy simulation study [30].

3. Results and discussion

3.1. Preparation of thermochromic VO₂:SiO₂ coatings with varying composition and thickness

To study the influence of composition and coating thickness on T_{vis} and ΔT_{sol} , we prepared coatings with three different VO₂:SiO₂ ratios and three different thicknesses. As a reference we prepared a pure VO₂ coating of approximately 100 nm thickness (Fig. 3d, Table 1). For varying composition we selected mixtures resulting in coatings with 40, 50 and 70% v/v VO₂ (Fig. 3a–c, Table 1). Therefore, we prepared mixtures of vanadyl oxalate solution and silica sol in appropriate molar ratios to lead to the desired composition, assuming standard densities for VO₂ and SiO₂ of 4.57 g cm⁻³ and 1.50 g cm⁻³, respectively. For all coatings with varying composition the overall solid content in the coating formulation and the withdrawal speed of the dip coater were kept constant to lead to coating thicknesses around 100 nm (Table 1). To investigate the influence of the coating thickness on the optical

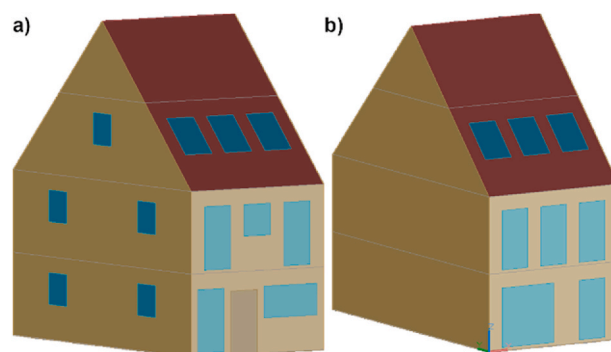


Fig. 2. Schematic depiction of residential building modeled in Energy Plus; a) north and east side, b) south and west side of the building.

properties we selected a 50% v/v composition. By adjusting the withdrawal speed of the dip coater at a constant solid content in the coating formulation, we prepared coatings with thickness of 47, 96 and 140 nm (Fig. 3e–g, Table 1). All coatings were prepared on Pilkington Optiwhite™ soda-lime glass with a pre-deposited silica barrier layer and were subjected to a 2-step curing procedure after dip coating. Here it was especially important to control the temperature and atmosphere during the curing procedures. The first curing step at 270 °C under air was needed to remove oxalate ligands and excess oxalic acid, and the second curing step under inert gas at 450 °C was required for the crystallization of VO₂ (TGA see supporting information, Fig. S3). The temperature for the crystallization step was hereby limited by the softening temperature of float glass of 470 °C. A qualitative analysis of the VO₂:SiO₂ coatings 1–3 showed optical homogeneity and no scattering of visible light. All coatings show low haze and glare values of < 1% with almost no reflection haze (Fig. S4). Furthermore, we qualitatively observed an increased tinting level with increasing VO₂ content and/or layer thickness, with the highest tinted coating 3 reaching almost the same tinting level as pure VO₂ (Fig. 3). A quantitative investigation of the optical properties including switching behavior is presented in section 3.3.

3.2. Structure and composition of VO₂:SiO₂ coatings

To investigate the influence of the addition of silica sol on the coating structure and the VO₂ domain size, we performed SEM and surface roughness analysis of all prepared coatings. Furthermore, to investigate a potential relationship between the surface structure and the mechanical robustness, we performed a pencil hardness test to investigate the scratch resistance of the thermochromic coatings. By combining SEM topography information from the Everhart-Thornley-Detector (ETD; Fig. 4) with material information from the Circular Backscatter Detector (CBS; Fig. S5), the surface structure of the coatings could be investigated in detail. We started by analyzing the pure VO₂ coating as reference. Via SEM we observed an incomplete and inhomogeneous surface coverage of the silica coated glass substrate with circular V poor regions, which are surrounded by V rich regions (Figs. 4a and S5a). Here partial dewetting of VO₂ occurred during thermal anneal, which led to a relatively rough surface structure. This inhomogeneous surface coverage

Table 1

Molar ratio and withdrawal speed for preparation of thermochromic coatings with varying VO₂:SiO₂ composition and thickness.

Coating	Coating composition VO ₂ :SiO ₂ [% v/v]	Coating formulation molar ratio VO ₂ :SiO ₂	Withdrawal speed [mm·s ⁻¹]	Coating thickness [nm]	Roughness [R _a]
Pure VO ₂	100:0	1:0	5	100	0.67
1	40:60	1.5:1	2.5	102	0.56
2a	50:50	2.2:1	1.3	47	0.41
2b	50:50	2.2:1	2	96	0.50
2c	50:50	2.2:1	4	140	1.14
3	70:30	5.1:1	2	96	0.69



Fig. 3. Thermochromic coatings at layer thickness around 100 nm and varying composition of a) 40% v/v (1), b) 50% v/v (2b), c) 70% v/v (3), d) 100% v/v VO_2 , and coatings at 50% v/v VO_2 at varying layer thickness of e) 47 nm (2a), f) 96 nm (2b), and g) 140 nm (2c).

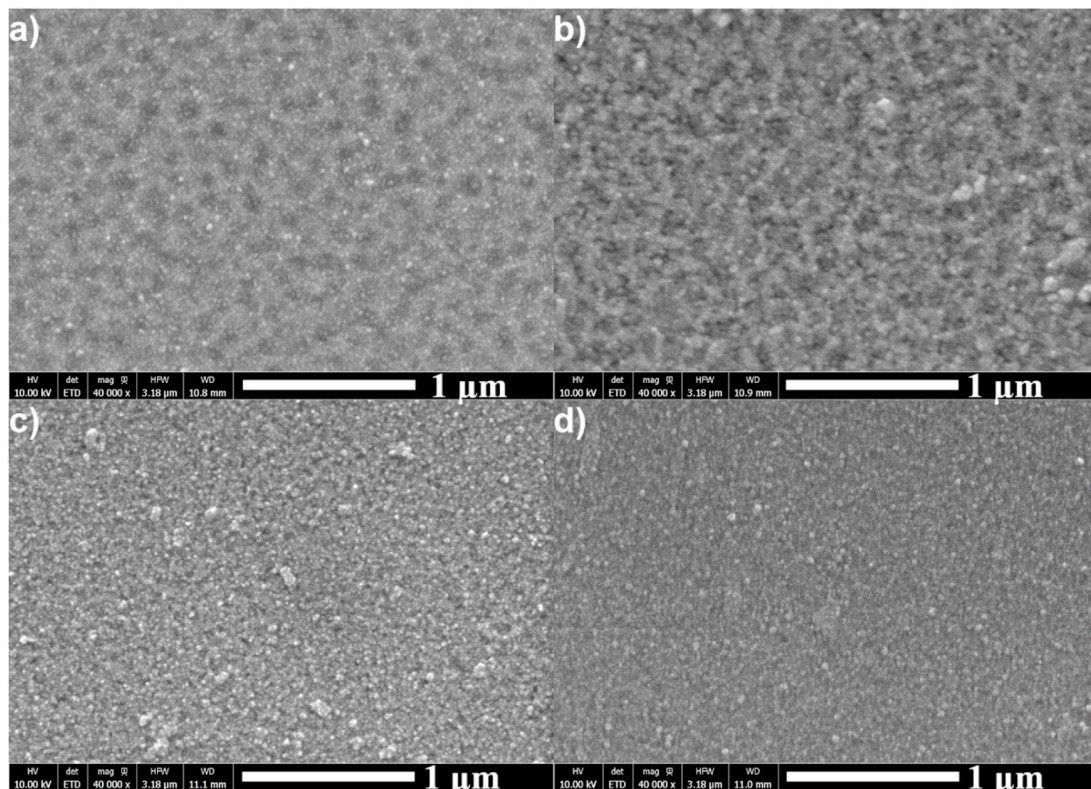


Fig. 4. SEM images of thermochromic coatings at layer thickness around 100 nm and varying composition of a) 100% v/v, b) 70% v/v (3), c) 50% v/v (2b), d) 40% v/v VO_2 (1).

and high surface roughness could be confirmed via optical profilometry measurements resulting in an R_q value of 0.67 (Table 1). Furthermore, it is reasonable to assume that this inhomogeneous, rough surface is sensitive to scratching, pollution, etc. This assumption could be confirmed by poor mechanical robustness represented by a pencil hardness $< 1\text{H}$.

By addition of silica to VO_2 , we intended to not only improve the optical properties of the thermochromic coating, as discussed in section 3.3, but also to improve the durability of the coating layer. Here the

intention was to create a silica matrix, which uniformly covers the substrate and embeds the VO_2 domains in a smooth coating layer. This way the matrix can protect the VO_2 domains against oxidation and the smooth surface structure enables sharp objects to slide over the surface without damaging it, hereby increasing scratch resistance. Analyzing VO_2 : SiO_2 coatings 1–3 via SEM, we observed that an increase in silica content led to an increased homogeneity in coating surface structure and V distribution in the coating layer (Fig. 4b–d, S5b–d). For coating 3 with

30% v/v SiO₂ still a relatively inhomogeneous surface structure was visible (Fig. 4b). The addition of SiO₂ led to a slight improvement when compared to the pure VO₂ coating. In coating 3 the de-wetting led to randomly structured and distributed domains with high and low coating thickness. Furthermore, a clear distinction between V rich and poor regions could be observed via back scattering images from the SEM CBS detector (Fig. S5b), showing an inhomogeneous distribution of V in the coating. Coating 3 still showed a similar roughness to the pure VO₂ coating with an *Rq* value of 0.69 (Table 1), but the addition of SiO₂ led to an increase in scratch resistance from < 1H to 4H. Further increase in SiO₂ content in the coating \geq 50% v/v for coating 1 and 2b yielded a homogeneous surface structure (Fig. 4c and d). Both coatings showed a relatively smooth surface with no clear distinction between thicker and thinner surface coverage of the substrate by the coating. This indicates a minimum content of SiO₂ in the coating mixture of 50% is needed to enable a smooth surface and homogeneous distribution of the V content in this coating matrix. A relatively homogeneous distribution of V in the coating matrix could be observed for coating 2b (Fig. S5c), which further improved for coating 1 with 60% v/v SiO₂, where almost no distinction between V rich and poor regions was possible via SEM (Fig. S5d). The homogeneous coating structure coincided with a low surface roughness and resulted in *Rq* values of 0.50 and 0.56 for coating 2b and 1, respectively (Table 1). Furthermore, an increased scratch resistance of 6H and 7H was observed for coatings 2b and 1, respectively.

Investigating coatings 2a-c with the same coating composition of 50% v/v VO₂ but different coating thickness of 47, 96 and 140 nm, respectively, led to no clear impact of coating thickness on surface structure and V distribution in the coating matrix. All three coatings showed a similar, relatively homogeneous surface structure and V distribution (Figs. S6 and S7). Nevertheless, an increase in coating thickness from 48 to 96 and 140 nm for coatings 2a-c, respectively, led to an increase in surface roughness (Table 1) and a decrease in scratch resistance from 7H to 6H and 5H for coatings 2a-c, respectively. All VO₂:SiO₂ coatings showed a better scratch resistance than sputtered silver coatings which are commonly used in energy-efficient low-e glazing [31].

Due to limitations in magnification, the exact VO₂ domain sizes in the coatings couldn't be analyzed via SEM, and the coating material was therefore further characterized via STEM-EDX (*vide infra*). To investigate

if the VO₂ domains form a directional pattern either in the pure VO₂ coating or in the VO₂:SiO₂ coatings, we performed UV-vis-NIR spectrophotometric analysis under a fixed angle with polarized light. Here anisotropy in the coatings should be visible by a change in light transmission for different polarization angles. All composite coatings 1-3, as well as the pure VO₂ coating, showed no change in light transmission upon change of polarization angle, indicating isotropic optical properties (Fig. S8).

For a more detailed analysis of phase separation between VO₂ and SiO₂ and to investigate the VO₂ domain size and distribution, we performed STEM-EDX analysis of scraped off coating material. We started by analyzing coating 2b in the three stages of preparation, i.e. as xerogel directly after dip coating, as amorphous material after the first annealing step and as partially crystalline material after the final anneal. Here we observed that within the xerogel stage we still have a homogeneous molecular mixture of VO₂ and SiO₂ without any phase separation (Fig. 5a). After the first annealing step we still see mainly a homogeneous mixture of VO₂ and SiO₂ with small elongated, curled needles on the surface (Fig. 5b-d). After the final anneal we observe that phase separation between VO₂ and SiO₂ has partly occurred with VO₂ mainly forming elongated domains of 100-200 nm in length and 20-50 nm in width (Fig. 5e-h). Here we can conclude that during the anneal process phase separation between VO₂ and SiO₂ starts on the surface of the coating during the first annealing step. In the second annealing step, the VO₂ crystals grow mainly in one direction resulting in elongated crystals of 100-200 nm length embedded in a SiO₂ matrix. Furthermore, we can conclude that not all vanadium species have been converted to crystalline monoclinic VO₂ and that other V-containing species are still partly present, homogeneously mixed in the SiO₂ matrix. Via DSC measurements of scraped off coating material we could show in a previous study that VO₂ (M) was the only crystalline material present in the coating with a degree of purity of approximately 54% with the remaining VO₂ content being amorphous [9].

After investigating the formation of VO₂ domains, we analyzed the impact of VO₂ content in the VO₂:SiO₂ coatings 1-3 on the final VO₂ domain size and distribution. Here we observed that not only the size and distribution but also the shape of the VO₂ domains was influenced by the VO₂ content. Coating 3 with a VO₂ content of 70% v/v showed mostly spherical or ellipsoidal VO₂ domains in a SiO₂ matrix (Fig. 6a).

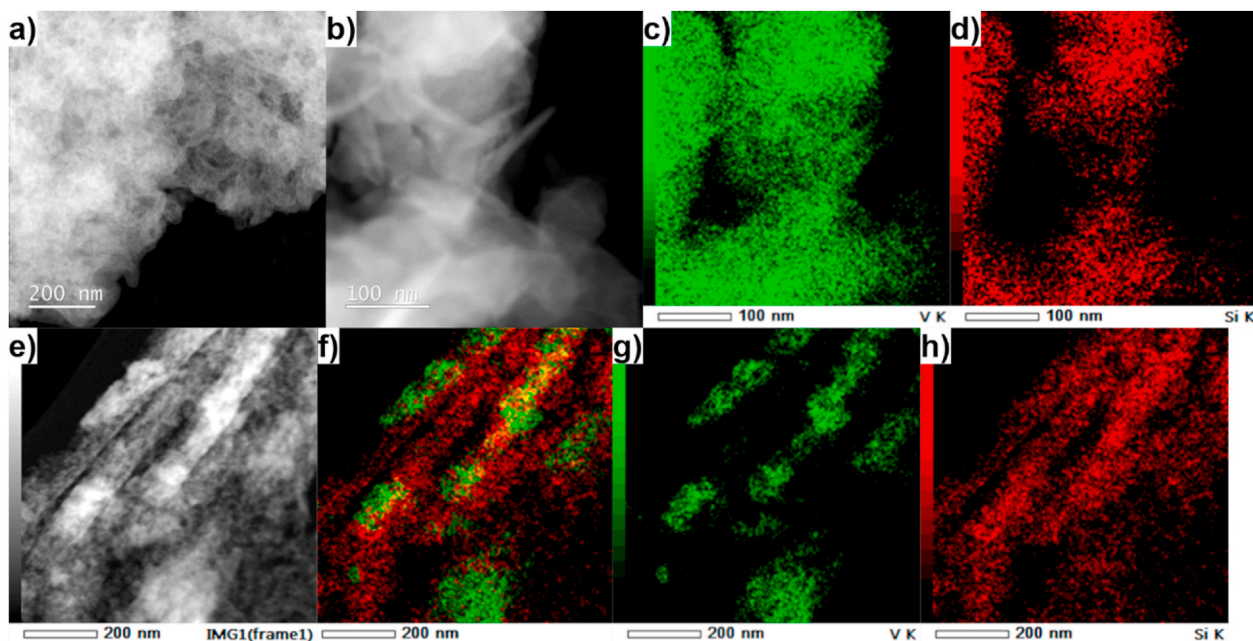


Fig. 5. STEM-EDX analysis of coating 2b as xerogel after dip coating (a), after the first annealing step (b-d) and in the final state (e-h) (EDX mapping: green = V, red = Si). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

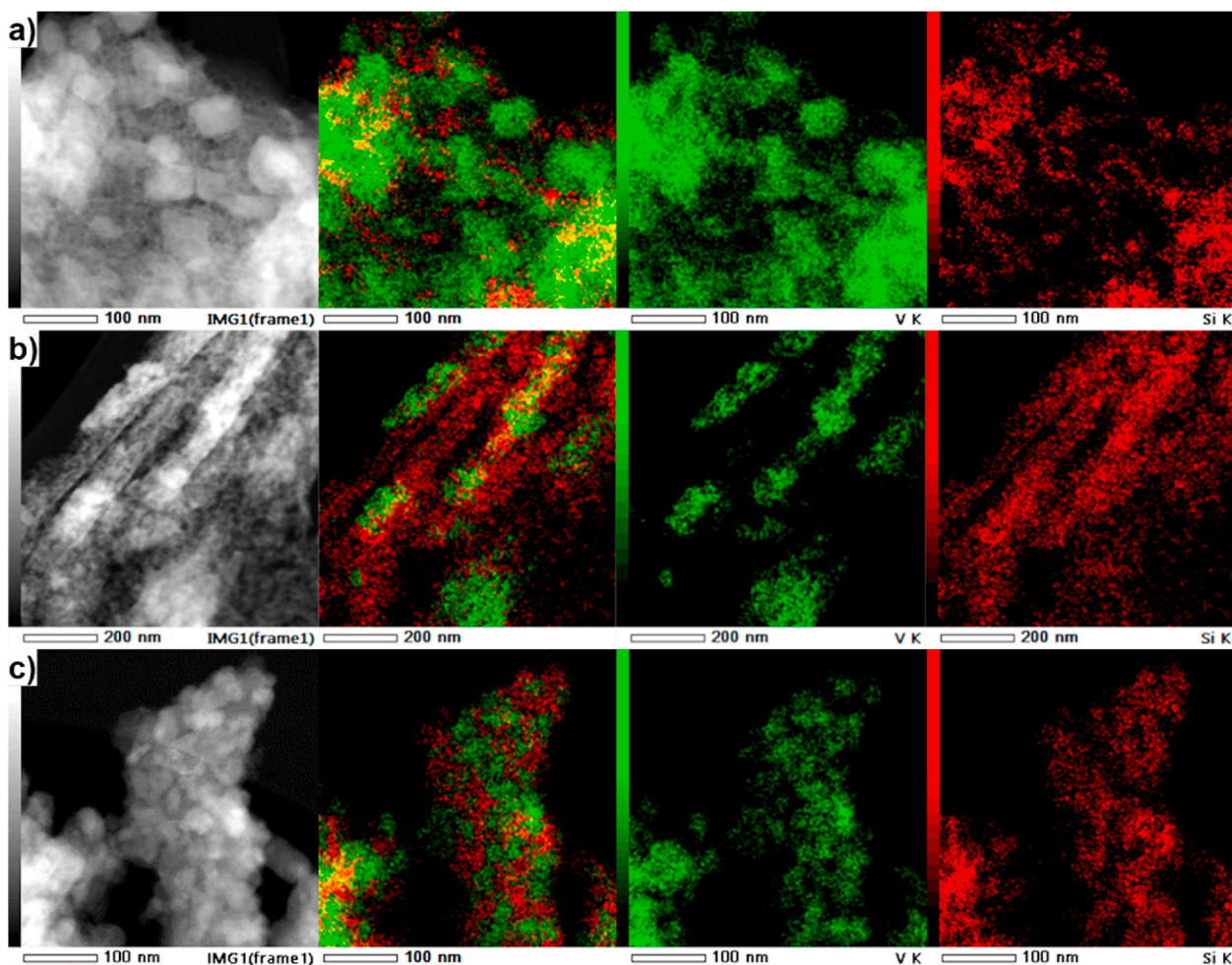


Fig. 6. STEM-EDX analysis of a) coating **3** with 70% v/v VO₂, b) coating **2b** with 50% v/v VO₂ and c) coating **1** with 40% v/v VO₂ (EDX mapping: green = V, red = Si). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

The domains showed a relatively small size distribution with an average diameter of 50 nm. Coating **2b** with a lower VO₂ content of 50% v/v showed mainly rod-shaped VO₂ domains with common lengths between 100 and 200 nm and thicknesses between 20 and 50 nm (Fig. 6b). In general the size distribution increased from coating **3** to **2b**. Coating **1** with the lowest VO₂ content of 40% v/v showed again mostly spherical or ellipsoidal VO₂ domains in a SiO₂ matrix (Fig. 6c). For this coating the VO₂ domains showed an average diameter of 20 nm with a small size distribution. In general it could be observed that the VO₂:SiO₂ ratio has a significant influence on VO₂ domain size and shape, with the smallest domain sizes realized at the lowest VO₂ content.

Furthermore, we analyzed the influence of coating thickness on VO₂ domain size, distribution and morphology in coatings **2a-c** with the same VO₂ content of 50% v/v. Here coating **2a** at 47 nm thickness showed mostly spherical or ellipsoidal VO₂ domains in a SiO₂ matrix (Fig. 7a). The domains showed a relatively small size distribution with an average diameter of 30 nm. Coating **2b** with an increased thickness of 96 nm showed mainly elongated VO₂ domains with common lengths between 100 and 200 nm and thicknesses between 20 and 50 nm (Fig. 7b). For coating **2c** with the largest coating thickness of 140 nm plate like VO₂ crystals were observed with a diameter of several hundred nm (Fig. 7c). As a general trend it was observed that with increasing coating thickness at a VO₂ content of 50% v/v, small spherical particles grow to elongated crystals and finally form plate-like VO₂ domains.

The in-depth STEM-EDX analysis of VO₂:SiO₂ coatings shows that phase separation between V and Si rich regions doesn't occur already after coating application and drying. In the xerogel stage, the vanadium

oxalate complex is still highly dispersed in the forming SiO₂ matrix. At temperatures at which the oxalate ligand starts decomposing, first VO₂ crystals are being formed, which further grow during the high temperature crystallization phase. In a 50% v/v VO₂ coating the crystal growth is faster than nucleation of new crystals, which leads to an increased crystal size with increased coating thickness. Here the increased amount of VO₂ leads to a crystal growth from small spheres, to elongated crystals and plate like particles. In a 40% v/v VO₂ coating, the crystal growth is slowed down by the increased dispersion. Here nucleation of new crystals is preferential, which leads to small spherical particles in the SiO₂ matrix. For composite coatings with an increased VO₂ content of 70% v/v, the low amount of SiO₂ is not able to form a complete matrix enabling free mobility of VO₂ during crystallization. Here VO₂ starts to crystallize locally and forms spherical crystals of medium size.

3.3. Influence of composition and coating thickness on thermochromic properties

Schläfer et al. reported optical simulations that predict that with increasing silica content VO₂ coatings at the same ΔT_{sol} values will show increasing T_{vis} [7]. Experimentally, they only provided a proof of concept and prepared coatings with high ΔT_{sol} values up to 16% at T_{vis} values of 46%. Furthermore, they investigated the influence of different silica sources (particles vs. sol). A detailed experimental investigation of the impact of coating thickness and composition on the optical properties was not within the scope of their study. To investigate this, we measured UV-vis-NIR transmission spectra for coatings with a thickness

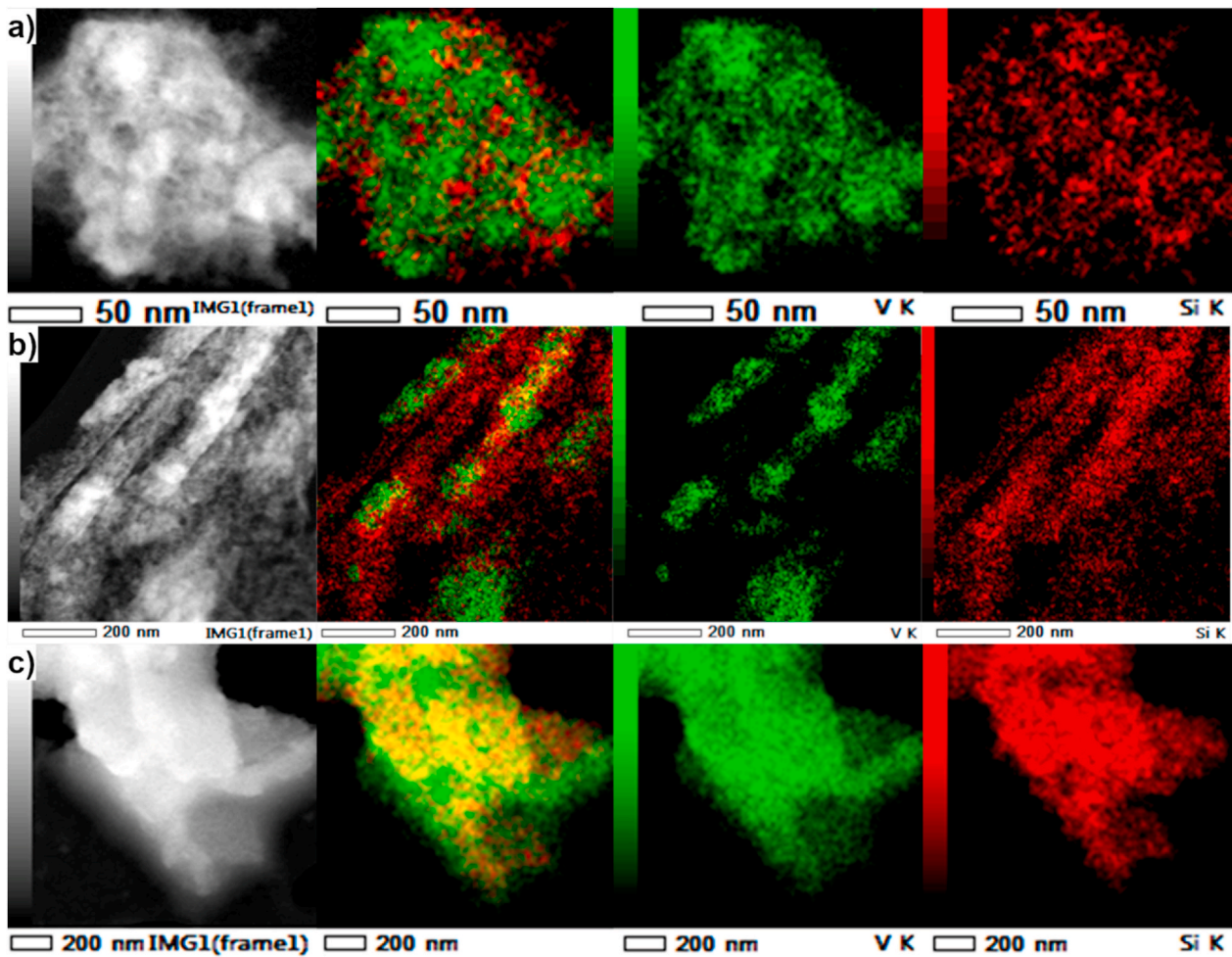


Fig. 7. STEM-EDX analysis of a) 47 nm thick coating 2a, b) 96 nm thick coating 2b and c) 140 nm thick coating 2c (EDX mapping: green = V, red = Si). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

around 100 nm and varying composition. The pure VO₂ coating with a thickness of 100 nm showed a T_{vis} in the cold state of 44% and a ΔT_{sol} of 5.6% (Fig. 8, Table 2). The low solar modulation results partly from the fact that the visible transmission increases to 49% when switching from cold to hot state. This reduces the overall solar modulation and is also not desired in the final application. When 30% v/v of SiO₂ was present in coating 3 with a thickness of 96 nm, the visible transmission slightly increased in the cold state in comparison to pure VO₂ (Fig. 8). Additionally, for the VO₂:SiO₂ coating 3 T_{vis} doesn't change upon phase

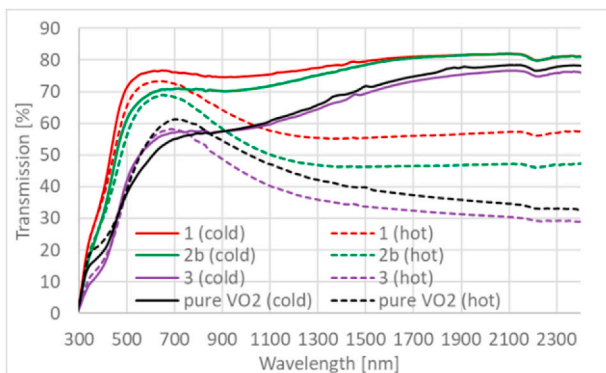


Fig. 8. UV-vis-NIR transmission spectra of VO₂:SiO₂ coatings 1–3 with varying compositions and coating thickness around 100 nm and pure VO₂ coating at similar coating thickness as reference.

Table 2

Optical properties of thermochromic coatings with varying VO₂:SiO₂ composition and thickness.

Coating	T_{vis} [%]	ΔT_{sol} [%]	T_{switch} [°C]	Hysteresis width [K]	Hysteresis gradient [K]	T_0 [°C]	ΔT_0 [K]
Pure VO ₂	44	5.6	84.8	44.8	6.8	81	30
1	74	10	81.4	36	8.3	78	26
2a	68	8.4	80.1	22.4	10.8	75	13
2b	66	11.8	90.8	28.2	10.9	85	17
2c	61	15.3	88.5	29	11.7	81	14
3	49	10.4	85.6	24.5	10.2	79	11

transition and the IR transmission in the hot state is slightly lower than for pure VO₂. This leads to a large increase in solar modulation in comparison to pure VO₂. Overall the VO₂:SiO₂ coating 3 with 70% v/v VO₂ and a thickness of 96 nm showed a T_{vis} of 49% and a ΔT_{sol} of 10.4% (Table 2). By increasing the silica content further to 50% v/v, the visible transmission could be increased even further. Coating 2b with a VO₂ content of 50% v/v and a thickness of 96 nm showed a largely increased T_{vis} in the cold state of 66% (Fig. 8, Table 2). Here also the IR transmission in the cold state is increased in comparison to the pure and the 70% v/v VO₂ coating. Additionally, a slight reduction in visible transmission to 63% upon switch from cold to hot state was visible and the IR transmission in the hot state was slightly higher than for pure VO₂. Still a large change in IR transmission occurred which leads to an overall ΔT_{sol}

of 11.8% (Table 2). Another change that was visible with addition of 50% v/v silica was a reduced absorption of violet (380–450 nm) and blue (450–485 nm) light, which resulted in less intense coloration of the coated glass plate (Figs. 3 and 8). Increasing the silica content above 50% v/v led to a further increase in visible transmission and a slight reduction in solar modulation. Coating 1 with a VO₂ content of 40% v/v and a thickness of 102 nm showed a T_{vis} of 74% and a ΔT_{sol} of 10.0% (Fig. 8, Table 2). The coating with the highest silica content showed a similar IR transmission in the cold state as the 50% v/v VO₂ coating 2b, but a slightly higher IR transmission in the hot state, which ultimately leads to a small reduction in solar modulation. Furthermore, the change in visible transmission upon phase transition increased slightly to $\Delta T_{vis} = -4\%$. Overall our results confirm the predictions by Schläfer et al. [7] that an increase in SiO₂ content in VO₂:SiO₂ composite coatings at similar coating thickness increased T_{vis} at a constant ΔT_{sol} .

Since the optical properties of the VO₂:SiO₂ coatings are also dependent on the coating thickness, we investigated the thermochromic properties of coatings with 50% v/v VO₂ and thicknesses between 48 and 140 nm. The VO₂:SiO₂ coating 2a with a thickness of 48 nm showed a high visible transmission of 68% in the cold as well as in the hot state (Fig. 9, Table 2). The change in transmission upon switch occurred exclusively in the IR region with a total solar modulation of 8.4% (Table 2). When doubling the coating thickness a slight decrease in visible transmission was observed and coincided with a significant increase in solar modulation. Coating 2b with a thickness of 96 nm showed a T_{vis} of 66% and a ΔT_{sol} of 11.8% (Fig. 9, Table 2). The change in IR transmission upon switch increased significantly in comparison to the thinner coating 2a. Additionally, a slight reduction in T_{vis} during phase transition was visible, which in combination led to the large increase in solar modulation. Further increase in coating thickness again led to a small reduction in visible transmission and a large increase in solar modulation. The VO₂:SiO₂ coating 2c with a thickness of 140 nm still showed a high T_{vis} in the cold state of 61%, which is only a reduction of 5% in comparison to coating 2b (Fig. 9, Table 2). But 2c showed an increased change in visible transmission upon switch to ΔT_{vis} of -5%. In combination with only a slight decrease in IR transmission in the cold state but a significant decrease in the hot state in comparison to 2b, this led to a large increase in ΔT_{sol} to 15% (Fig. 9, Table 2). The coating 2c showed the highest solar modulation of all coatings prepared in this study. Overall the results show that for increasing coating thickness at constant coating composition a large increase in ΔT_{sol} can be achieved with only slightly reducing T_{vis} , which can have significant impact on the potential for application in energy-efficient glazing.

For application of VO₂ coatings in energy-efficient windows it is important to combine high T_{vis} with high ΔT_{sol} values. For all 6 categories of thermochromic VO₂ coatings and films mentioned in the introduction we selected 5 reported coatings/films with the highest combination of T_{vis} and ΔT_{sol} and compared them to the coatings

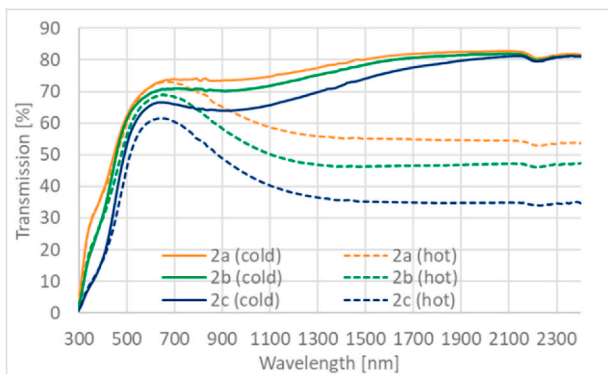


Fig. 9. UV-vis-NIR transmission spectra of VO₂:SiO₂ coatings 2a-c with 50% v/v VO₂ and varying coating thickness.

reported in this study (Fig. 10). It can be seen that especially coatings 1, 2b and 2c exceed almost all other reported high performing VO₂ coatings and are similar in performance to nanocomposite films. The few VO₂ coatings with comparable performance are either not commercially feasible, due to the use of hydrothermal growth of coatings on glass [14], freeze drying [15] or the use of complex (hollow) core shell particles [16,17], or combine high solar modulation with a low visible transmission of $T_{vis} < 55\%$ [20–23]. In this comparison, we can show that with our VO₂:SiO₂ coatings we have reached the performance of nanocomposite films at T_{vis} values $> 60\%$. For application purposes these coatings show high potential as coated glass in energy-efficient windows, whereas nanocomposite films are suited for laminated glass only.

The very good performance of the VO₂:SiO₂ coatings in this study can be explained by the distribution of VO₂ in a dielectric medium and by the purity and crystallinity of the functional VO₂ in the coatings. A measure for purity and crystallinity of VO₂ is the transition enthalpy in J g⁻¹. To analyze the VO₂ domains in our composite coatings, we used data of DSC measurements of scraped off coating material that we reported previously [9]. Here the thermodynamic switching temperature (T_0) for the VO₂ (M) → VO₂ (R) and VO₂ (R) → VO₂ (M) phase transition was 72 °C and 62 °C, respectively. The switching enthalpy of the analyzed coating material, consisting of 50% v/v VO₂, was 22.3 J g⁻¹ corresponding to a degree of purity of the crystalline VO₂ (M) phase of approx. 54%.

To investigate the influence of silica content and coating thickness on the phase transition, we measured temperature dependent transmission spectra of all 5 VO₂:SiO₂ coatings and the pure VO₂ coating as reference. The coated glass plates were heated up in small temperature steps to above the switching temperature and were subsequently slowly cooled down. Hysteresis plots were prepared for all coatings and the hysteresis width and gradient were analyzed (Fig. 11a and b, Figs. S9–S13, Table 2). The pure VO₂ coating showed a switching temperature (T_{switch}) from cold to hot state at 84.4 °C and a very broad hysteresis width of 44.8 K with a small gradient upon heating of 6.8 K (Fig. S9, Table 2). The thermodynamic switching temperature (T_0) for the pure VO₂ coatings was at 81 °C for the insulator to metal transition (IMT), with a difference (ΔT_0) between T_0 for IMT and metal insulator transition (MIT) of 30 K. The high T_{switch} and broad hysteresis width are due to the structure of the pure VO₂ coating, which consists of inhomogeneous VO₂ clusters of various sizes and an inhomogeneous coverage of the substrate (Fig. 4a). Addition of 30% v/v of silica in coating 3 led to a significant reduction in hysteresis width to 24.5 K and only a minor increase in transition gradient to 10.2 K (Fig. S10, Table 2). Here the silica matrix led to a more homogeneous surface coverage and decreased size distribution of VO₂ domains (Fig. 4b). Furthermore, it was observed that the average

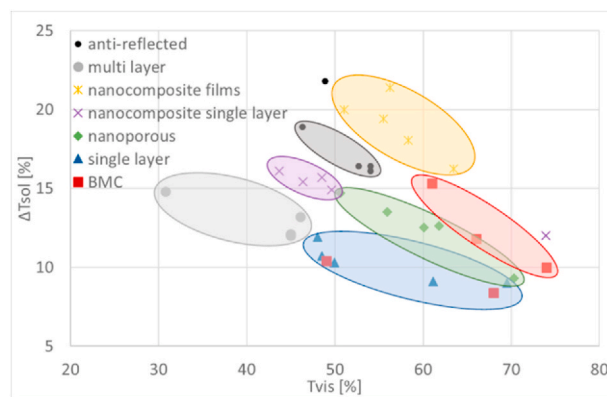


Fig. 10. Performance comparison between best performing thermochromic coatings/films and our reported VO₂:SiO₂ coatings (red squares). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

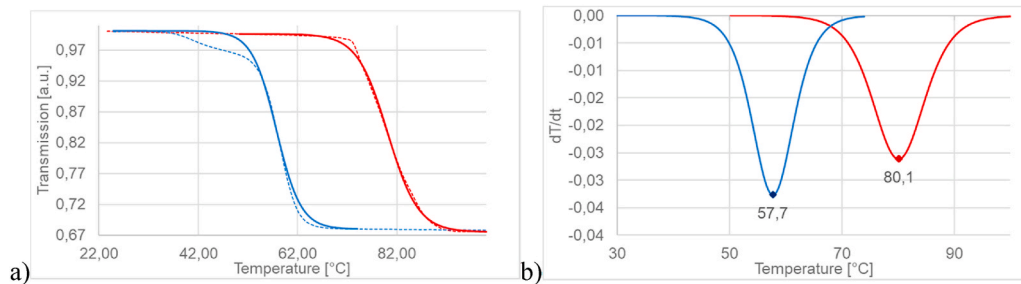


Fig. 11. a) Measured (dotted) and fitted (full line) relative transmission at 1600 nm of VO₂:SiO₂ coating **2a** as function of temperature upon heating (red) and cooling (blue). b) First derivative of hysteresis plot with indication of T_{switch} , hysteresis width and gradient. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

VO₂ domain size was reduced to 50 nm (Fig. 6a). The smaller domain size and therefore larger surface area, coinciding with a small size distribution, leads to a narrowing of the hysteresis width. T_{switch} was not impacted by addition of 30% v/v silica, whereas T_0 was slightly reduced to 78 °C and ΔT_0 was reduced to 11 K. Further increase in silica content with similar coating thickness led to an increase in hysteresis width to 28.2 K and 36.0 K for coatings **2b** and **1**, respectively, without significant change in hysteresis gradient (Figs. S11 and S12, Table 2). It was observed that for coating **2b** the VO₂ domains had a rod-like shape with a length between 100 and 200 nm and a width between 20 and 50 nm (Fig. 6b). Additionally, the size distribution increased slightly. The changes in VO₂ domain size, size distribution and shape negatively influence the hysteresis gradient. For coating **1**, spherical VO₂ domains with an average diameter of 20 nm and a small size distribution were observed (Fig. 6c). No clear trend for the change in T_{switch} was observed since it first increased to 90.8 °C for coating **2b** with 50% v/v silica and then was reduced significantly to 81.4 °C for coating **1** with 60% v/v silica. The same observation was made for T_0 , which was increased to 85 °C for coating **2b** and reduced to 78 °C for coating **1**. Furthermore, ΔT_0 increased with increasing silica content to 17 and 26 K for coating **2a** and **1**, respectively. Via the thermodynamic switching temperature values, it can be concluded that for spherical VO₂ domains a reduction in domain size leads to a reduction in switching temperature. The increased surface to volume ratio is likely to increase the number of surface defects and therefore the nucleation sites for the phase transition [32]. Here the deviation of coating **2b** can be explained by the change in VO₂ domain geometry from spherical to elongated. Investigating the influence of coating thickness on the transition temperature, as well as on the hysteresis gradient and width, we observed that at thicknesses >100 nm no major change occurred (Fig. S13, Table 2). T_{switch} , T_0 and ΔT_0 were slightly reduced and transition hysteresis and gradient were slightly increased by increasing the coating thickness from 96 to 140 nm at constant VO₂ content for coatings **2b** and **c**, respectively. The coating **2c** with a thickness of 140 nm showed plate-like VO₂ domains with average diameters of several hundred nm (Fig. 7c). Here the structure of the VO₂ domains has a similar impact on the phase transition as for coating **2b** with elongated crystals. Only very thin coatings of 48 nm showed a slightly lower hysteresis width of 22.4 K and a significantly lower T_{switch} and T_0 of 80.1 and 75 °C, respectively (Fig. 11, Table 2). For coating **2a**, small spherical VO₂ domains with average diameter of 30 nm and a small size distribution were observed (Fig. 7a). The similar size and shape of VO₂ domains for coating **1** and coating **2a** at different coating thickness and composition resulted in similar T_{switch} , T_0 and hysteresis gradient values. Therefore we can conclude that the IMT is mainly influenced by the VO₂ domain size and shape. ΔT_0 and transition hysteresis deviated significantly, since the MIT for the higher diluted coating **1** started at a much lower T_0 . Here the matrix shows a significant impact on the phase transition from rutile to monoclinic, which increases the difference in T_0 between IMT and MIT. In general it can be concluded that the VO₂ domain size and shape, as well as the matrix can have a major influence on the switching behavior. Small, spherical VO₂

domains with a small size distribution lead to a low IMT temperature and a small hysteresis gradient. Overall introducing VO₂ domains in a silica matrix increases the transition temperature for the MIT and reduces the hysteresis width. Here, increased dilution slightly reduces the MIT temperature and increases the hysteresis width.

3.4. Long-term stability of VO₂:SiO₂ coatings

To select relevant durability and lifetime tests for VO₂:SiO₂ coatings, we have to consider where the coating will be placed in the final application as a smart window and what kind of conditions the coating is exposed to during production and use of an IGU. Coatings for energy-efficient glazing are usually applied on side 2 (Fig. 1) of an IGU. In that position they show the best performance in reducing solar or radiator heat transmission, and are also shielded from the outside environment and therefore better protected against degradation. For thermochromic coatings the positioning on side 2 has an additional importance, since the switch of the material has to be based on the temperature of the exterior glass pane. The atmosphere in the gap of high performance IGUs commonly consists of argon, which means that the coating is not in contact with oxygen. The only requirement for an energy-efficient coating placed on side 2 of an IGU is to withstand UV radiation according to EN 1096-3 [33]. To analyze UV stability similar to EN 1096-3, we exposed a coated glass plate with a 60 nm thick VO₂:SiO₂ coating with 70% v/v VO₂ to UV radiation for more than 1000 h. All testing conditions, such as setup and UV source, were set to comply with the definitions in the standard taking into account limitations due to sample size. To analyze potential degradation, we measured UV-vis-NIR spectroscopy before and after 139, 726 and 1220 h exposure to UV radiation (Fig. S14). No change in optical properties in the hot and the cold state could be observed over the whole testing period. Therefore, we can conclude that the coating withstands UV radiation according to EN 1096-3 and passes the critical requirements for energy-efficient coatings in windows.

Since the coating is protected in the final application within an IGU with argon atmosphere, UV stability is the only relevant stability criterion for the thermochromic coating in the finished smart window. Nevertheless, after coating application until assembly of the IGU several months can pass, where the coating will be subjected to standard air atmosphere. Therefore, it is important for the industrial processing of the coated glass that it can withstand several months in contact with air under standard conditions (20 °C, 40% humidity) without deterioration of the optical properties. To investigate the degradation behavior of the VO₂:SiO₂ coatings in air we exposed a coated glass plate with a 50 nm thick coating with 70% v/v VO₂ to the standard environment in our lab of 20 °C and 40% humidity and measured UV-vis-NIR spectroscopy for a duration of 11 months. Within this period no deterioration of the optical properties was observed in the hot and the cold state (Fig. S15). Therefore it can be concluded that the stability of the VO₂:SiO₂ coatings is sufficient for industrial processing into IGUs.

3.5. Energy savings potential of VO₂:SiO₂ coatings

To evaluate the energy savings potential of the high performance VO₂:SiO₂ coatings, we performed building energy simulations. Here we designed a typical Dutch building and used Beek as a location for the simulations. Since > 60% of Dutch residential buildings are either duplex or terraced houses [34], we selected a building with one side without contact to the outside environment. Furthermore, we designed wall insulation to reach an Rc value (thermal resistance of the composite) of 4.7 m² K W⁻¹, in agreement with current Dutch regulations. Additionally, we selected a size of 130 m² living space and 35 m² of windows, which are common values for a 4 person household in the Netherlands (Figs. 2 and S1). In the simulation we used VO₂:SiO₂ coatings 1, 2b and 2c, which show the highest combination of T_{vis} and ΔT_{sol} in a double glazing unit combined with a standard low-e coating. The annual energy consumption for a building equipped with either one of these three smart windows or a low-e glazing was compared to clear double glazing to calculate annual energy and cost savings (Fig. 12). With these simulations we could show that all three analyzed smart windows can significantly reduce annual energy consumption and lead to energy savings between 23 and 24%, which is a relative improvement of 10% over low-e glazing. The smart window with VO₂:SiO₂ coating 1 shows the highest energy savings of 23.8%. Here the high T_{vis} of 66% with a cold state G value of 0.52 and a U value of 1.0 W m⁻² K⁻¹ of the smart window, leads to a reduction of annual heating demand of 16.4%, which is the highest energy savings for heating of all analyzed smart windows. Since heating demand for the simulated building in Beek equals > 70% of the total energy costs, the window with the highest energy savings for heating has also the highest impact on total energy savings. The smart windows with VO₂:SiO₂ coatings 2b and 2c show even higher reduction in annual cooling demand with up to 71% savings due to hot state G values of 0.46 and 0.41, respectively. But due to lower energy savings for heating of 14.0% and 10.2%, respectively, the overall energy savings are with 23.7% and 23% lower than for the smart window with coating 1. Taking into account the average cost price for electricity and gas in the Netherlands the annual cost savings for the three smart windows can be calculated [35]. Here an individual household can save up to 470 € per year in comparison to the use of double clear glass.

4. Conclusions

In conclusion, we developed an industrially applicable procedure to synthesize high performance thermochromic coatings on glass via liquid

coating deposition with subsequent thermal curing. The combination of VO₂ and SiO₂ in a composite coating made from sol-gel mixtures leads to small homogeneously distributed VO₂ domains in a silica matrix. During thermal anneal of the xerocoat, phase separation yielding V rich and Si rich regions occurs. We investigated this phase separation in detail and concluded that the crystal growth is dependent on the composition (V:Si ratio) and thickness. Due to the high crystallinity and small particle size of the VO₂ domains in the SiO₂ matrix unrivalled optical properties are achieved. The combination of $T_{vis} > 60\%$ and $\Delta T_{sol} \geq 10\%$ of the composite coatings is exceeding comparable coatings reported to date, and makes these coatings suitable for application in smart windows. We showed that the balance between T_{vis} and ΔT_{sol} can be influenced by the VO₂:SiO₂ ratio and the coating thickness, enabling the customization of optical properties for different climate zones or building types for optimum energy savings. The different VO₂ domain sizes and shapes and the silica matrix have a major impact on the phase transition. We demonstrated that small, spherical VO₂ domains at a low dilution in a silica matrix lead to optimized transition temperature and hysteresis gradient and width. The silica matrix additionally leads to a smooth coating surface and encapsulates the VO₂ domains. This leads to an increased stability of the composite coatings in comparison to pure VO₂ coatings. Here we demonstrated that the composite coatings withstand stability tests relevant for energy-efficient coatings in IGUs. Additionally, we showed that with a scratch resistance $\geq 4H$ and a stability in ambient environment of at least 11 months, the coatings are suitable for industrial processing into IGUs. With building energy simulations on a typical Dutch building, we demonstrated that the use of smart windows with our thermochromic coatings can lead to energy savings of 24%, which equals to annual cost savings of up to 470 € per household.

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CRedit authorship contribution statement

Cindy P.K. Yeung: Validation, Formal analysis, Investigation, Visualization. **Roberto Habets:** Validation, Investigation. **Luc**

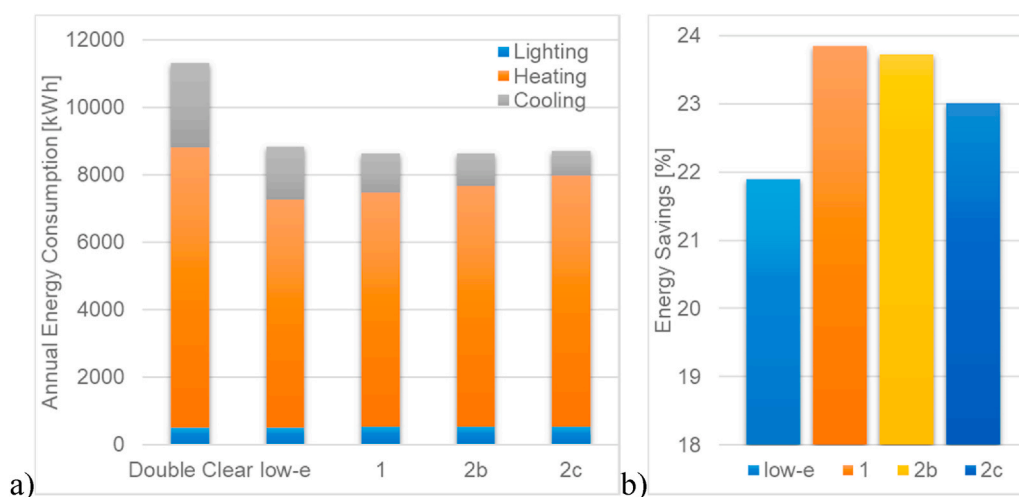


Fig. 12. a) Annual energy consumption of simulated residential building equipped with clear double glazing, low-e glazing or one of the three smart windows and b) annual energy savings of the smart windows over double clear glass and energy savings of low-e glazing as comparison.

Leufkens: Validation, Investigation, Visualization. **Fallon Colberts:** Validation, Investigation. **Kathleen Stout:** Conceptualization, Validation. **Marcel Verheijen:** Investigation, Visualization. **Zeger Vroon:** Conceptualization, Validation. **Daniel Mann:** Conceptualization, Methodology, Validation, Writing – original draft, Writing – review & editing, Visualization. **Pascal Buskens:** Conceptualization, Writing – review & editing, Supervision.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.solmat.2021.111238>.

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